Safety Data Sheet Date of Issue/re-issue: 01.03.2016

User declaration:- I have read and understood this Safety Data Sheet

Name:-						
		Signature	<u> </u>			Date
Name:-						
		Signature	Date			
1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER						
Company Name						
ECP Limited ECD						
Address:		39 Woodside Ave, Northcote, Auckland , New Zealand				
Emergency Tel: NZ		Tel +64 9 480 4386		FAX +64 9 480 4385		
0800154666						
Product	L-Ascorbic ac	cid AR		Code		14201
CAS#	HSNO#	UN#	DG Clas	DG Class/es Packir		ing group #

Also known as Vitamin C

50-81-7

2. Hazard Identification

Not classified as hazardous according to the Approved Criteria for Classifying Hazardous Substances

na

na

na

Not classified as dangerous goods

GHS classification

None

3. Composition/information on ingredients

Chemical Solid

Characterization

Ingredients Name

L-Ascorbic acid. 100% CAS 50-81-7

4. First-aid measures

Remove victim to fresh air. Seek medical advice if effects persist. If breathing has stopped, apply artificial respiration.

Inhalation

Rinse mouth thoroughly with water immediately. Give plenty of water to drink.

Seek medical attention in severe cases, or if large amounts ingested. NEVER give anything by mouth if victim is rapidly losing

consciousness, is unconscious or is convulsing. Do not induce vomiting.

Ingestion

Wash with plenty of soap and water. If irritation occurs seek medical advice.

Remove contaminated

clothing Wash clothing before reuse. Seek medical attention.

Skin

Irrigate with copious quantity of water for 15 minutes. Seek medical assistance if symptoms persist.

Eye contact

Remove contact lenses.

First Aid Facilities Maintain eyewash fountain and drench facilities in work area.

Treat symptomatically.

Adverse effects on colour vision. Ascorbic acid should be given with care to patients with hyperoxaluria.

Advice to Doctor

For advice, contact a Poisons Information Centre New Zealand 0800 764 766) or a doctor.

Other Information

5. Fire-fighting measures

Not classified as hazardous

Hazards from Acrid smoke and irritating fumes, carbon monoxide and carbon dioxide.

Combustion

Products

Small fire: Use dry chemical, CO2, water spray or foam.

Large fire: Use water spray, fog or foam.

If safe to do so, move undamaged containers from the fire area. Cool containers with flooding quantities

of water until well after the fire is out.

Specific Methods

May burn but do not ignite readily. Runoff may pollute waterways. Fire may produce irritating, poisonous

and/or corrosive fumes. Containers may explode when heated.

Specific hazards arising from the chemical

Decomposition ca. 183 °C; 190 - 192 °C. Temp.

Precautions in Wear SCBA and structural firefighter's uniform.

connection with Fire

6. Accidental release measures

Eliminate all ignition sources (no smoking, flares, sparks or flame) within at least 15m. Do NOT touch or

walk through this product. Stop leak if safe to do so. Prevent entry into waterways, drains, confined

areas. Prevent dust cloud. Use clean non-sparking tools to collect material and place it into

loosely-covered plastic containers for later disposal.

SEEK EXPERT ADVICE ON HANDLING AND DISPOSAL.

Spills & Disposal

Do not breathe dust. Do not breathe fumes, vapour, gas. Avoid inhalation, contact with skin, eyes and

clothing.

Personal

Precautions

7. Handling and storage

Avoid ingestion and inhalation of vapours, or dusts. Avoid contact with eyes, skin, and clothing. Avoid

prolonged or repeated exposure. Minimize dust generation and accumulation. Keep container tightly

closed. Use with adequate ventilation. If ingested, seek medical advice immediately and show the

container or the label. Wash thoroughly after handling. Remove contaminated clothing and wash before

reuse. Light sensitive. Protect against physical damage and light. Keep away from heat and all sources

of ignition. Ground all equipment containing material. Keep away from incompatibles such as oxidizing

agents and strong bases. Containers of this material may be hazardous when empty since they retain

product residues (dust, solids); observe all warnings and precautions listed for the product.

Precautions for Safe

Handling

Store in tightly closed containers, in a cool, dry, well-ventilated area away from incompatible substances.

Separated from strong oxidants, strong bases. Solutions of ascorbic acid are rapidly oxidized in air and

in alkaline media; the drug should be protected from air and light. Store in light-resistant containers.

Store away from direct sunlight. Store protected from moisture. Store under an

inert atmosphere. Protect

against physical damage. Isolate from any source of heat or ignition. Containers of this material may be

hazardous when empty since they retain product residues (dust, solids); observe all warnings and

precautions listed for the product.

Conditions for safe

storage, including

any

incompatabilities

Corrosiveness Attacks aluminium, copper, copper alloys, iron and zinc.

Storage Store at room temperature (15 to 25 °C recommended).

Temperatures

Unsuitable Materials Aluminium, copper, copper alloys, iron and zinc.

8. Exposure controls/personal protection

A time weighted average (TWA) concentration for an 8 hour day, and 5 day week has not been established

Other Exposure

Information

In industrial situations maintain the concentrations values below the TWA. This may be achieved by

process modification, use of local exhaust ventilation, capturing substances at the source, or other methods.

Appropriate engineering controls

Where ventilation is not adequate, respiratory protection may be required. Avoid breathing dust, vapours

or mists. Respiratory protection should comply with AS 1716 - Respiratory Protective Devices

Respiratory Protective Devices. Filter capacity and respirator type depends on exposure levels. In event of emergency or

planned entry into unknown concentrations a positive pressure, full-facepiece SCBA should be used. If

respiratory protection is required, institute a complete respiratory protection program including selection,

fit testing, training, maintenance and inspection.

Respiratory Protection

Eye Protection The use of a face shield, chemical goggles or safety glasses with side shield protection as appropriate.

Hand protection should comply with AS 2161, Occupational protective gloves -

Selection, use and

maintenance. Recommendation: Plastic or rubber gloves.

Hand Protection

Clean clothing or protective clothing should be worn, preferably with an apron.

Clothing for protection

against chemicals should comply with AS 3765 Clothing for Protection Against Hazardous Chemicals.

Body Protection

Always wash hands before smoking, eating or using the toilet. Wash contaminated clothing and other

protective equipment before storing or re-using.

Hygiene Measures

9. Physical and chemical properties

Form Solid

Appearance White to a very pale yellow crystalline solid, or colourless crystals.

Odour Odourless.

Decomposition ca. 183 °C; 190 - 192 °C.

Temperature

Melting Point ca. 183 °C; 190 - 192 °C (decomposes).

Solubility in Water Soluble (1g/3mL at 20 °C).

Slightly soluble in alcohol, glycerol and propylene glycol. Insoluble in ether, chloroform, benzene,

petroleum ether, oils and fats.

Solubility in Organic Solvents

Specific Gravity 1.65 - 1.70

Solubility in Fat Insoluble.

pH 2.1 - 2.6 (5% aqueous solution)

Vapour Pressure 7.9179 Pa @ 192 °C

Volatile Component 0 %vol @ 21 °C

Partition Coefficient: log P (o/w): -2.15

n-octanol/water

Flash Point 99 °C

Flammability Combustible.

Auto-Ignition 640 °C

Temperature

Flammable Limits - 10% by Volume (g.cu. ft.)

Lower

Flammable Limits - 20% by Volume (g.cu. ft.)

Upper

Explosion May form an explosive organic dust cloud with air.

Properties

Molecular Weight 176.13

Oxidising Properties The substance is a strong reducing agent and reacts with oxidants.

Other Information Taste: Pleasant, sharp acidic taste.

10. Stability and reactivity

Stable at room temperature in closed containers under normal storage and handling conditions. Stable

to air when dry; aqueous solutions are rapidly oxidized by air.

Chemical Stability

Conditions to Avoid Heat, ignition sources, light, air, moisture, dust generation and incompatible materials.

Strong oxidizing agents, alkalies, aluminium, iron, copper, copper alloys, zinc, metal ions, water, acids,

sodium nitrate, alkali hydroxides, sodium salicylate, sodium nitrite, theobromine and methenamine.

Incompatible Materials

Hazardous Carbon monoxide and carbon dioxide.

Decomposition Products

Reactive with oxidizing agents.

Air and light sensitive.

Aqueous solutions are rapidly oxidized by air, accelerated by alkalies, iron, copper.

Possibility of

hazardous reactions

Hazardous Will not occur.

Polymerization

11. Toxicological Information

LD50 (rat): 11900 mg/kg; LD50 (mouse): 3367 mg/kg.

Acute Toxicity - Oral

Ingestion of small amounts during normal industrial handling is a low hazard.

Ingestion of large amounts

may cause gastrointestinal irritation, gastrointestinal disturbances (nausea, vomiting and diarrhoea),

hypermotility, acidification of the urine which may cause precipitation of cystine and oxalate stones in the

urinary tract and may cause renal failure coordination, somnolence, eyes (lacrimation), blood (anaemia),

a disruption of psychological functioning resulting in decreased reaction times and psychomotor

coordination. Increases iron absorption and, thus, large doses may be dangerous in persons with

haemochromatosis, thalassaemia, or sideroblastic anaemia. Persons with erythrocyte/G6PD defifiency

may develop mild haemolysis. High doses taken during pregnancy may cause scurvy in infants when

born.

Ingestion

Inhalation Irritates the respiratory tract. May cause coughing and a sore throat.

May cause mild to moderate irritation and redness. When in solution forms a strong acid which may be

irritating to skin.

Skin

May cause mild to moderate irritation, redness and pain. When in solution forms a strong acid which

may be irritating to eyes.

Eye

Carcinogenicity Not listed in the IARC Monographs.

Chronic ingestion of ascorbic acid can change the pH of the saliva so that calcium is lost from tooth

enamel leading to dental enamel erosion. Prolonged or repeated ingestion may affect the blood/bone

marrow and metabolism. Chronic ingestion of large doses may cause gastrointestinal disturbances

including nausea and diarrhoea, urinary effects involving urine acidification, oxalate and uric

crystallization in the bladder and kidney, and decreased reaction times and psychomotor coordination.

Chronic Effects

Serious eye Eye irritation test (rabbit): Slight irritation.

damage/irritation

Skin Skin irritation test (rabbit): No irritation.

corrosion/irritation

12. Ecological information

No ecological problems are to be expected when the product is handled and used with due care and attention.

Ecological Information

When introduced properly, no impairments in the function of waste-water-treatment plants are to be expected.

Ecotoxicity

Biologic degradation: Readily biodegradable. BOD 48 % of ThOD /5 d (closed bottle test); BOD 65 % of ThOD /28 d (closed bottle test).

Persistence and degradability

Mobility Distribution: log P(oct) -2.15.

Bioaccumulative No bioaccumulation is to be expected (log P(o/w < 1)).

Potential

Onchorhynchus mykiss LC50: 1020 mg/l /96 h acid;

L.idus LC50: 33000 mg/l /48 h.

Acute Toxicity - Fish

13. Disposal considerations

Disposal Dispose of according to relevant local, state and federal government regulations.

Considerations

14. Transport information

15. Regulatory information

Poisons Schedule Not Scheduled

16. Disclaimer

The information above is believed to be accurate and represents the best information currently available to us. However, the information is not a guarantee expressed or implied, with respect to such information, and we assume no liability resulting from its use. Anyone using the chemical described here should ensure that he or she has the appropriate training and has the expertise and any equipment required for safe handling. If clarification or further information is required, please contact ECP Ltd or refer to the official handler of dangerous goods within your own company. The user should also make their own investigations to determine the suitability of the product for their particular purposes. In no event shall the company be liable for any claims, losses, or damages of any third party or for lost profits or any special, indirect, incidental, consequential, or exemplary damages howsoever arising, even if the company has been advised of the possibility of such damages.

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